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In chemistry, a colloid is a mixture in which one substance of microscopically dispersed insoluble particles is suspended throughout another substance. Sometimes the dispersed substance alone is called the colloid; the term colloidal suspension refers unambiguously to the overall mixture (although a narrower sense of the word suspension is distinguished from colloids by larger particle size).

Colloid - Wikipedia

Solids in liquids (sols): properties – kinetic, optical and electrical; stability of colloids, protective action, Hardy-Schulze law, gold number.

Chemistry - Chhatrapati Shahu Ji Maharaj University, Kanpur

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Flocculation, in the field of chemistry, is a process wherein colloids come out of suspension in the form of floc or flake, either spontaneously or due to the addition of a clarifying agent. The action differs from precipitation in that, prior to flocculation, colloids are merely suspended in a liquid and not actually dissolved in a solution. In the flocculated system, there is no formation of a ...

Flocculation - Wikipedia

Product Reports. Skip to the Lab Analysis Summary. There are three distinctly different types of silver that are labeled and sold on the market as “colloidal silver”; they are ionic silver, silver protein, and true colloidal silver. Consumers seeking true colloidal silver are often at a disadvantage because each of these products represents themselves as colloidal silver.

Silver Colloids: Colloidal Silver Product Reports

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